

Code: CE1T3, CS1T3, EC1T4, IT1T3

**I B. Tech - I Semester – Regular/Supplementary Examinations  
November 2017**

**ENGINEERING CHEMISTRY  
(Common for CE, CSE, ECE & IT)**

Duration: 3 hours

Max. Marks: 70

**PART – A**

Answer *all* the questions. All questions carry equal marks

11 x 2 = 22 M

1.

- a) Define hardness of water? What are the causes of hardness?
- b) Which one is most widely used chlorination process?
- c) How the plastics are classified?
- d) Write the name and structure of bullet proof plastic.
- e) Give the applications of nonmaterial in food industries.
- f) Write few characteristics of fullerenes.
- g) What are the main functions of the paint?
- h) Steel screw in a brass marine hardware corrodes. Explain.
- i) Give the importance of pilling bed worth rule.
- j) What are liquid crystals?
- k) What are high TC super conductors?

## PART – B

Answer any **THREE** questions. All questions carry equal marks.

3 x 16 = 48 M

- 2.a) One liter of water from an underground reservoir in Tirupathi Town in Andhra Pradesh showed the following analysis for its contents:  $\text{Mg}(\text{HCO}_3)_2 = 42 \text{ mg}$ ;  
 $\text{Ca}(\text{HCO}_3)_2 = 146 \text{ mg}$ ;  $\text{CaCl}_2 = 71 \text{ mg}$ ;  $\text{NaOH} = 40 \text{ mg}$ ;  
 $\text{MgSO}_4 = 48 \text{ mg}$ ; organic impurities = 100 mg; Calculate temporary, permanent and total hardness of this sample of 10,000 liters of water. 8 M
- b) How is the exhausted resin regenerated in an ion exchanger? What are the merits and demerits of ion exchange method? 8 M
- 3.a) Mention the compounding materials used in plastics indicating their function and give one example for each? 8 M
- b) Discuss the preparation, properties and applications of Bakelite. 8 M
- 4.a) What is green Chemistry? What are the twelve principles of green chemistry? 8 M

- b) Green Chemistry is sustainable chemistry- Explain the statement. **4 M**
- c) What do you mean by nano materials? Explain different categories of nano materials with suitable examples. **4 M**
- 5.a) Define pitting corrosion. State the conditions necessary for pitting corrosion to occur. State the hazards and controls associated with pitting corrosion. **8 M**
- b) What do you mean by sacrificial anodic protection? Explain with suitable examples. **8 M**
- 6.a)What are liquid crystals? What are the advantages of liquid crystal display? **5 M**
- b) Discuss the effect of temperature on semi conductors. **3 M**
- c) What do you understand by a semi conductor? Discuss some properties of semi conductors. **8 M**